

# **Zn<sub>3</sub>P<sub>2</sub>-based solar cell devices**

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A retrospective survey of the current status of Zn<sub>3</sub>P<sub>2</sub>-based solar cells is presented. The paper includes a short review of the fundamental properties of Zn<sub>3</sub>P<sub>2</sub>, the discussion of some technological problems in the devices and the theoretical estimations of the parameters of Zn<sub>3</sub>P<sub>2</sub>-based solar cells. Finally, the best achievements in the prototype devices area are thoroughly considered, comprising the opaque-metal-grid and transparent-metal-film Schottky-type devices and the Zn<sub>3</sub>P<sub>2</sub>-based heterojunction-type ones.

## **1. Introduction**

Zinc phosphide (Zn<sub>3</sub>P<sub>2</sub>) has become one of the promising semiconductors as the absorbing-generating material for applications in photovoltaic solar energy conversion. Both the values of the direct energy bandgap and the diffusion length of minority carriers (see below) promote reasonably high energy conversion efficiency. Furthermore, both Zn and P are abundant and relatively inexpensive, which is perhaps even the equally important feature. Prototype Zn<sub>3</sub>P<sub>2</sub>-based solar cell devices have been recently fabricated and tested, strengthening a need to promote further investigations of both the fundamental properties of Zn<sub>3</sub>P<sub>2</sub> and the electronic performance of Zn<sub>3</sub>P<sub>2</sub>-based devices.

The successful development of any photoelectric device can be traced through a number of distinct stages. The first step is to identify a semiconductor for potentially effective application in photoconversion systems. To identify such a semiconductor means to find out the energy location and shape of the fundamental absorption edge; to estimate minority carrier diffusion length and material resistivity; to establish the thermal and chemical stabilities of the semiconductor and device; and, finally, to figure out the mineralogical abundance of constituents and cost of their extraction and refinement. Following that there come studies of photon collection efficiency (PCE), quantum efficiency of photo-generation processes (QE), carrier collection efficiency (CCE), and then parameters of prototype devices at various configurations. The next stage is the production and development of the prototype devices leading finally to a large-scale production technology. Considering the development of the Zn<sub>3</sub>P<sub>2</sub>-based solar cells, both the first and the second stages have been successfully completed.

It is a goal of this paper to discuss shortly the basic properties of  $\text{Zn}_3\text{P}_2$  and to present the current status of the  $\text{Zn}_3\text{P}_2$ -based prototype devices for broad-range photoconverter applications.

## 2. Optical properties of $\text{Zn}_3\text{P}_2$

Any rational design of a photovoltaic solar cell device rests on the knowledge of energy position of the fundamental absorption edge; the value of the absorption coefficient above it up to 3.5 eV approximately; and the energy factor of reflectance within the same energy range.

The smallest energy bandgap of  $\text{Zn}_3\text{P}_2$  has been ascribed to indirect transitions with onset at 1.315 eV and the fundamental absorption edge to direct transitions at the gamma point with onset at 1.51 eV, both values at room temperature [1]–[3]. Absorption coefficient of  $\text{Zn}_3\text{P}_2$  has been found to be relatively high above the fundamental absorption edge, i.e. higher than  $10^4 \text{ cm}^{-1}$ . The energy factor of reflectance for normal-incident unpolarized light reflected from a single  $\text{Zn}_3\text{P}_2$  crystal is discussed in [1]. The results of reflectivity for polarized light and the real

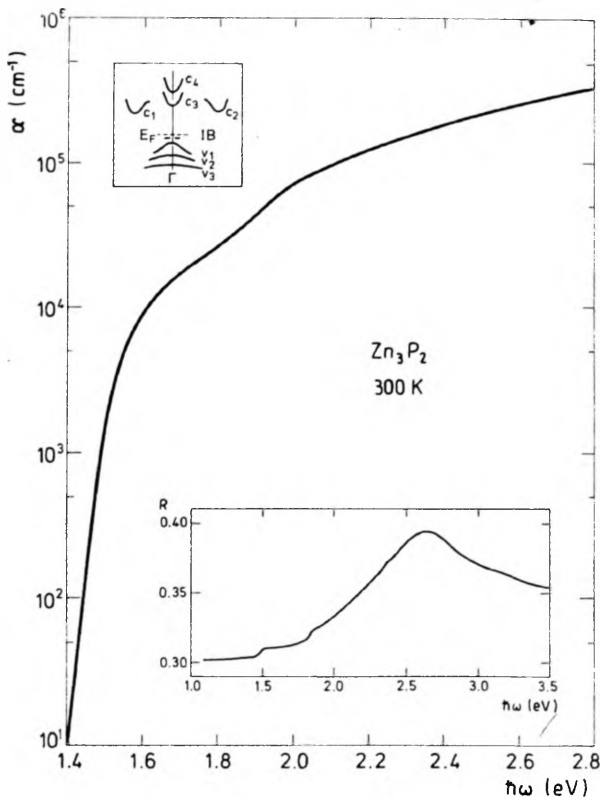


Fig. 1. Experimental data of  $\text{Zn}_3\text{P}_2$  absorption coefficient and reflectivity factor (lower insert) near the fundamental absorption edge. Upper insert shows the schematic band diagram of  $\text{Zn}_3\text{P}_2$  from [2]

and imaginary parts of complex dielectric constant of oriented Zn<sub>3</sub>P<sub>2</sub> single crystals are presented in [4]. Energy factor of reflectance for thin Zn<sub>3</sub>P<sub>2</sub> films is shown in [5]. Standard absorption coefficient and reflectivity spectra of Zn<sub>3</sub>P<sub>2</sub> (for unpolarized light) are shown in Fig. 1 together with the schematic band diagram proposed after [2].

Lattice modes of Zn<sub>3</sub>P<sub>2</sub> were studied by infrared absorption in [6]. The principal absorption bands lay in 20–25 micrometer waverange and numerous smaller bands were seen and attributed to multiphonon processes. Infrared absorption of single Zn<sub>3</sub>P<sub>2</sub> crystals were also studied in [7]. The multiphonon absorption within 0.06–0.11 eV and impurity absorption within 0.5–1.4 eV range have been found. Particularly, the impurity and/or impurity-like levels have been identified at energies of 0.17, 0.26, 0.36, 0.65, 0.87 and 1.28 eV, at 300 K.

### 3. Electronic properties of Zn<sub>3</sub>P<sub>2</sub>

Both majority and minority carrier transport properties and their recombination in the bulk and on the surface of semiconductor are of obvious importance to the successful design of any solar cell.

All samples of Zn<sub>3</sub>P<sub>2</sub> measured so far have shown p-type conductivity over the temperature range of 80–380 K. The room-temperature resistivity has been usually between 10 and 200 ohm × cm, corresponding to the average value of hole concentration of 10<sup>21</sup>–10<sup>22</sup> cm<sup>-3</sup> and hole mobility of (10–30) 10<sup>-4</sup> m<sup>2</sup>V<sup>-1</sup> sec<sup>-1</sup>. The resistivity of as-grown samples has apparently depended on preparation conditions, ambient atmosphere, and thermal history of the sample. Ionized-impurity scattering at low temperatures and acoustic-phonon scattering at high temperatures have been found as the dominating mechanisms [8], [9]. The surface recombination velocity has been found to dominate over the bulk one. However, the presence of grain boundaries has not appeared to be detrimental to the bulk recombination. The experimentally observed dependence of acceptor concentration on phosphorus pressure [10] suggested phosphorus interstitials to be acting as acceptors. Analysis of the resistivity and Hall-effect measurements in the 4.2–300 K temperature range shown in [11] has resulted in suggestion of two (at least) acceptor levels with energy of 0.035 and 0.6 eV above the valence band.

A minority carrier diffusion length has a great impact upon the solar cell performance and configuration. The diffusion length of electrons in Zn<sub>3</sub>P<sub>2</sub> has been estimated by means of several methods to be of the order of a few micrometers. The spectral photoresponse measurements [12], [13], electron-beam induced current method [14], and laser-spot scanning method [15] have been reported.

### 4. Other properties of Zn<sub>3</sub>P<sub>2</sub>

The Zn<sub>3</sub>P<sub>2</sub> lattice has (up to approx. 1053 K) tetragonal symmetry belonging to the  $D_{4h}^{15}$  space group. The dimensions of the unit cell are  $a = b = 0.8097$  nm and

$c = 1.145$  nm, and this cell contains 40 atoms. At about 1053 K the tetragonal phase transforms to the cubic one with lattice constant 0.582 nm, and this cell contains 10 atoms. Some information on crystal structure, preparation and the thermodynamic properties of  $\text{Zn}_3\text{P}_2$  has been given in [16]. A brief discussion of the relationship of bonding to the crystal structure has been published in [5].

Among many semiconductors being investigated actually for application as solar cell devices, only a few meet the requirements of abundance and low cost of fabrication. The power potential of these materials (in GW) based on the known world reserves and a 4-micrometer thick semiconductor layer is very large (more than  $10^6$  GW) for poly-Si, a-Si and  $\text{Zn}_3\text{P}_2$ ; and equal to  $10^5$  GW for CdS, 400 GW for GaAs, 200 GW for CdTe and 12 GW for InP [17]. Therefore, the practically possible number of semiconductors seems to be limited to the first four ones, with  $\text{Zn}_3\text{P}_2$  included.

## 5. Problems with $\text{Zn}_3\text{P}_2$

### 5.1. Doping

Concerning the conductivity type, there are two related phenomena found in  $\text{Zn}_3\text{P}_2$ : only p-type conductivity has been observed and there was impossible to obtain n-type conductivity material in any known way. Both phenomena are, in a general sense, related to defect chemistry in  $\text{Zn}_3\text{P}_2$ . Although the attempt to understand the p-type conductivity mentioned above has been made in [10], this effort ended up with indication of phosphorus interstitials as a source of p-type conductivity, with no detailed analysis of the on-n-type doping problem. The change from p-type to n-type conductivity suggested in [14] to be a result of heating the Mg- $\text{Zn}_3\text{P}_2$  Schottky devices seems rather to be caused by the formation of some compounds (e.g.,  $\text{Mg}_3\text{P}_2$ ) and/or dipoles on the surface (if chemical reaction model of barrier formation is applicable). Magnesium seems to be the only metal that is capable of providing this type of surface reactions (see also below for further discussion). On the other hand, crystal structure of  $\text{Zn}_3\text{P}_2$  is highly complex (40 atoms per unit cell) and defective. In addition to phosphorus interstitials, the presence of Zn vacancies has been firmly established [16]. Therefore, self-compensation effect is likely to occur in  $\text{Zn}_3\text{P}_2$ . It refers to compensation of any donor impurity due to native defects generated in the crystal lattice. This effect is currently under thorough examinations.

### 5.2. Surface properties

$\text{Zn}_3\text{P}_2$ -surface properties and metal- $\text{Zn}_3\text{P}_2$  interface reactions are relatively uninvestigated; furthermore they are unusually interesting for the ohmic contact and barrier contact fabrications (see also discussion on doping problem above). More than ten metals were used to prepare the contacts with  $\text{Zn}_3\text{P}_2$  [18]–[20] and the contact-barrier heights for the metal used did not show simple Schottky-type dependence on metal work-function. Instead, the results correlated quite well with the

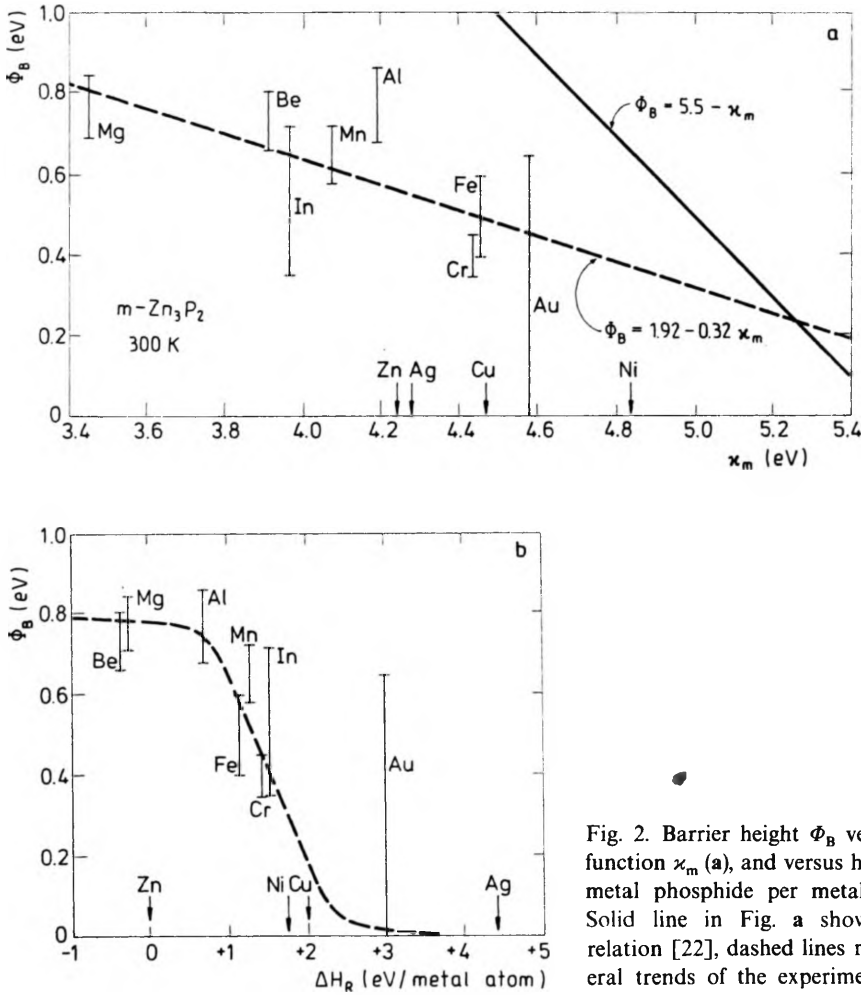


Fig. 2. Barrier height  $\phi_B$  versus metal work function  $\chi_m$  (a), and versus heat of reaction of metal phosphide per metal atom  $\Delta H_R$  (b). Solid line in Fig. a shows the Schottky relation [22], dashed lines represent the general trends of the experimental results

chemical reaction model proposed in [21] with Mg provided the highest barrier obtainable.

The results of measurements (taken from [19] and from our lab) are presented in Figs. 2a, b. The barrier height  $\phi_B$  measured by various methods (current-voltage and capacitance-voltage characteristics and internal over-the-barrier photoemission spectra) for twelve metal contacts is shown in Fig. 2a along with the metal work function. The great scatter of results for In and Au particularly should be noticed. Furthermore, visible difference between the simple Schottky relation [22] shown in Fig. 2a as a solid line and the general trend of the experimental results should be underlined. The experimental data can be approximated (according to the model including the surface states effect [23]) linearly and this approximation is shown as dashed line in Fig. 2a. Surface state density can be estimated [23] from this dependence. Its value (for the data from Fig. 2a) is of the order of  $10^{17} \text{m}^{-2} \text{eV}^{-1}$ ,

approximately\*, which indicates a relatively high influence of the surface states upon the electrical properties of the metal-Zn<sub>3</sub>P<sub>2</sub> contacts – within the frame of the model considered.

The barrier height versus heat of reaction of metal phosphide,  $\Delta H_R$  (per metal atom) of twelve metals with Zn<sub>3</sub>P<sub>2</sub> is shown in Fig. 2b. The results are taken from [19] and from our measurements. The better correlation between the general shape of dependence of  $\Phi_B$  vs.  $\Delta H_R$  (dashed line) and the experimental results emerges from this figure then from Fig. 2a. It adds strength to the concept of the chemical bonding at the interface metal-zinc phosphide. An important feature of this concept is the occurrence of the maximum value of  $\Phi_B$ ; in this case it is equal to 0.8 eV approximately. Therefore, the metal-Zn<sub>3</sub>P<sub>2</sub> junctions under AM1 condition of illumination will not show the open-circuit voltage greater than 0.5 V approximately.

Preparation of a low-resistivity ohmic contact with Zn<sub>3</sub>P<sub>2</sub> is still an apparent problem and Ag seems to be the best choice according to [19], [20]; see also Figs. 2a, b. On the other hand, the long-term stability of Zn<sub>3</sub>P<sub>2</sub>-based devices are mostly dependent on the surface/interface behaviour. The stability is of a great importance in Mg-Zn<sub>3</sub>P<sub>2</sub> Schottky-type devices especially, for Mg<sub>3</sub>P<sub>2</sub> (if created) being unstable.

### 5.3. Photoresponse origin

Thorough studies on photovoltage response of metal-Zn<sub>3</sub>P<sub>2</sub> devices [18], [20], [24], [29], [30] showed the negligible effect of recombination centers within the Zn<sub>3</sub>P<sub>2</sub> energy gap on the measured spectral characteristics, although the presence of these centers is highly probable (see Sect. 3). They could result in enhanced tunneling and thus in deteriorated conversion performance. In contrast, the study suggests relatively strong impact of surface/interface recombination. Also, the study revealed the space-resolved photovoltage effect, i.e., the strong influence of the light spot position in between the contacts on the sign and value of the photovoltage response. That effect is, in the first approximation: i) related to the two-type-carrier excitations (electrons from impurity levels and/or indirect band-to-band transitions and those from the direct ones); ii) influenced by the photostatic effect in high-resistivity samples, and iii) caused by the high surface recombination, as well. This problem has recently been discussed thoroughly in [30].

## 6. Zn<sub>3</sub>P<sub>2</sub>-based solar cell devices

### 6.1. Theoretical estimations

The computations have been made of photon collection efficiency (PCE), carrier collection efficiency (CCE), short-circuit current ( $j_{sc}$ ) and total conversion efficiency (TCE), of the prototype Zn<sub>3</sub>P<sub>2</sub>-based solar cell devices [25]–[27]. Results of these calculations, together with discussion of quantum efficiency (QE) of photo-genera-

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\* The value of  $8 \times 10^{16} \text{m}^{-2} \text{eV}^{-1}$  has recently been obtained [28].

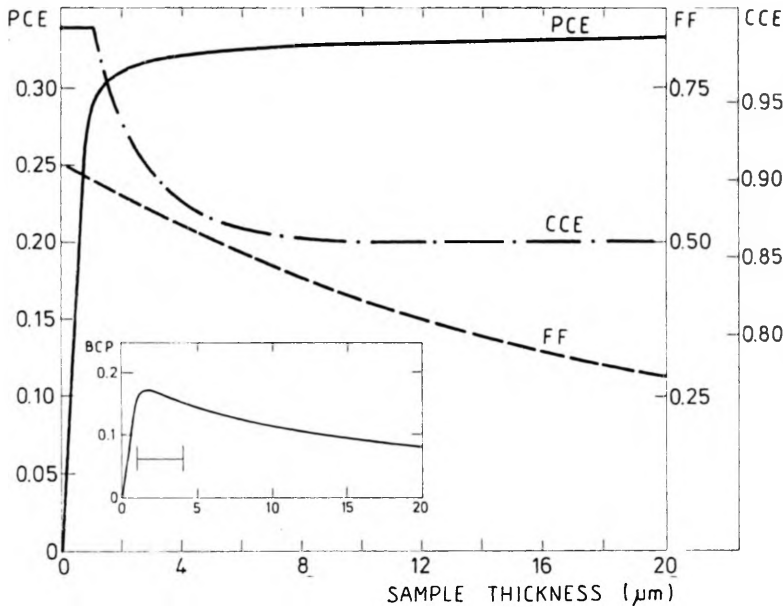


Fig. 3. Photon-collection and carrier-collection efficiencies versus sample thickness, computed for bare-surface front-wall configuration Zn<sub>3</sub>P<sub>2</sub>-based solar cell. Fill factor computed by taking into account the contact resistivity equal to 10 ohms is also shown as a function of thickness. BCP factor (see the text) versus sample thickness is shown as an insert

tion processes in Zn<sub>3</sub>P<sub>2</sub> are extensively analyzed in [25], [26]. The fill-factor (FF) of the prototype devices has been computed in [27] by taking into account the generation-recombination processes in the field-in-built region as a dominating mechanism in carrier transport through the junctions and the series resistivity of the devices as caused by the Zn<sub>3</sub>P<sub>2</sub> resistivity only, with contact resistivity being negligible.

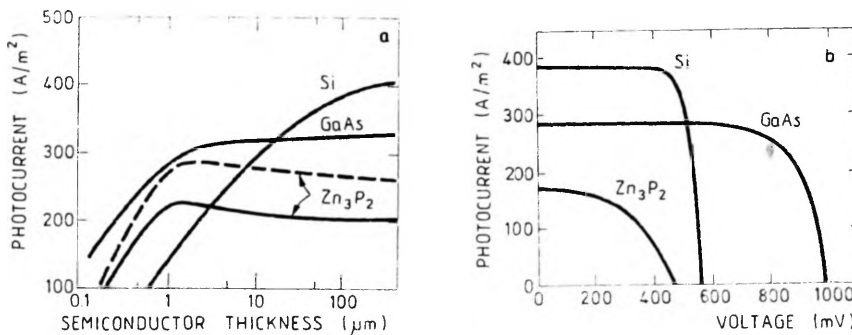
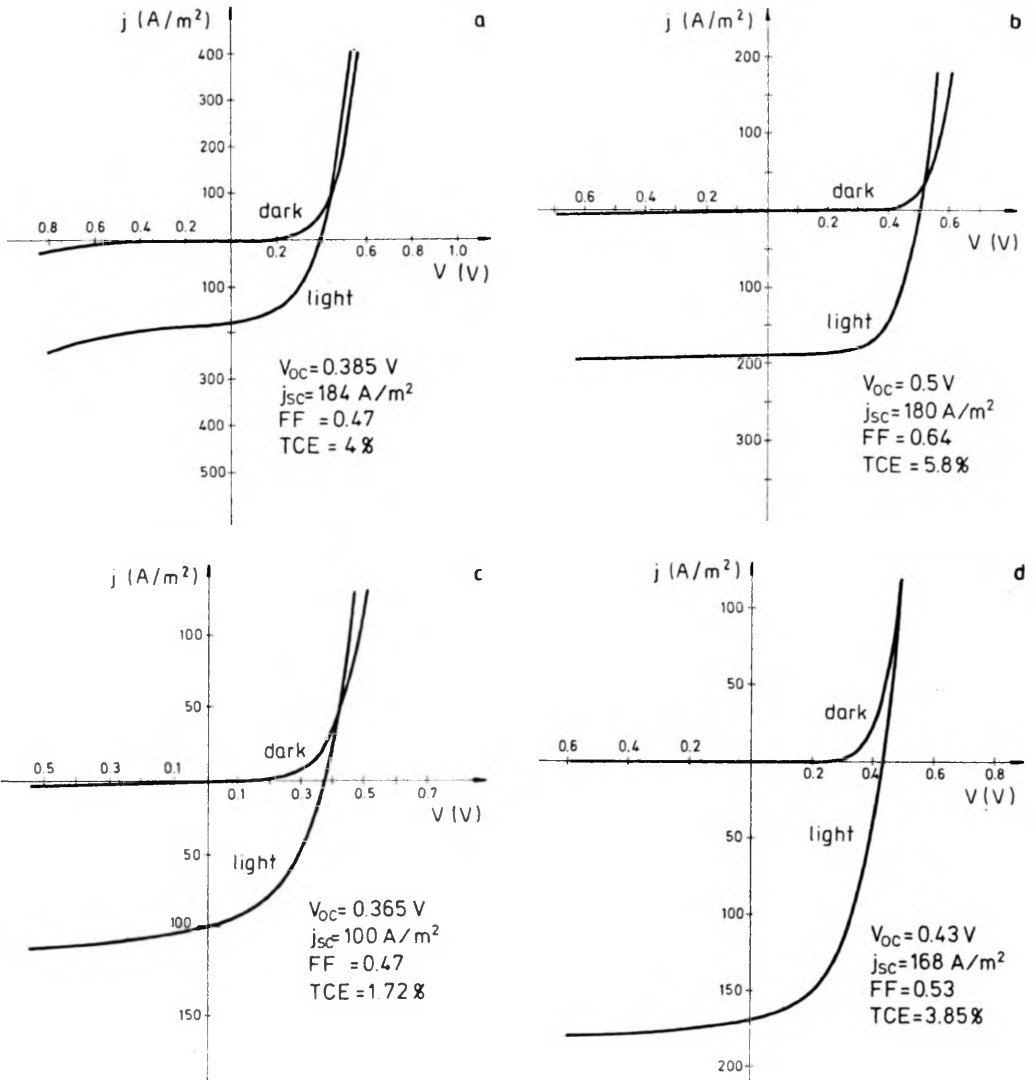


Fig. 4. Peak value of photocurrent (short-circuit current density) computed in a function of the semiconductor base thickness under AM1 illumination condition (a), and the photocurrent-versus-voltage experimental characteristics (b) taken for Si and GaAs from [32] and for Zn<sub>3</sub>P<sub>2</sub> from [27]



All the fundamental parameters of the solar cell devices are thickness-dependent. The plots of PCE, CCE and FF versus  $Zn_3P_2$  thickness are shown in Fig. 3. Their product, called here the best-compromized parameters (BCP) factor, is shown there as an insert. The BCP-versus-thickness plot clearly indicates that the most appropriate thickness of  $Zn_3P_2$  sample for solar energy conversion is within the range of 1.5–4 micrometers.

Loos figures and output data (taken from [27]) of four theoretically considered  $Zn_3P_2$ -based devices are listed in the Table, including the opaque-metal grid (OMG) and transparent-metal film (TMF) Schottky-type metal-semiconductor junctions, p-n homojunction and p-n heterojunction. Results for bare  $Zn_3P_2$  sample are also shown for comparison. The computations have been made under AM1.5 conditions of illumination (density of power equal to  $832 \text{ Wm}^{-2}$ ).



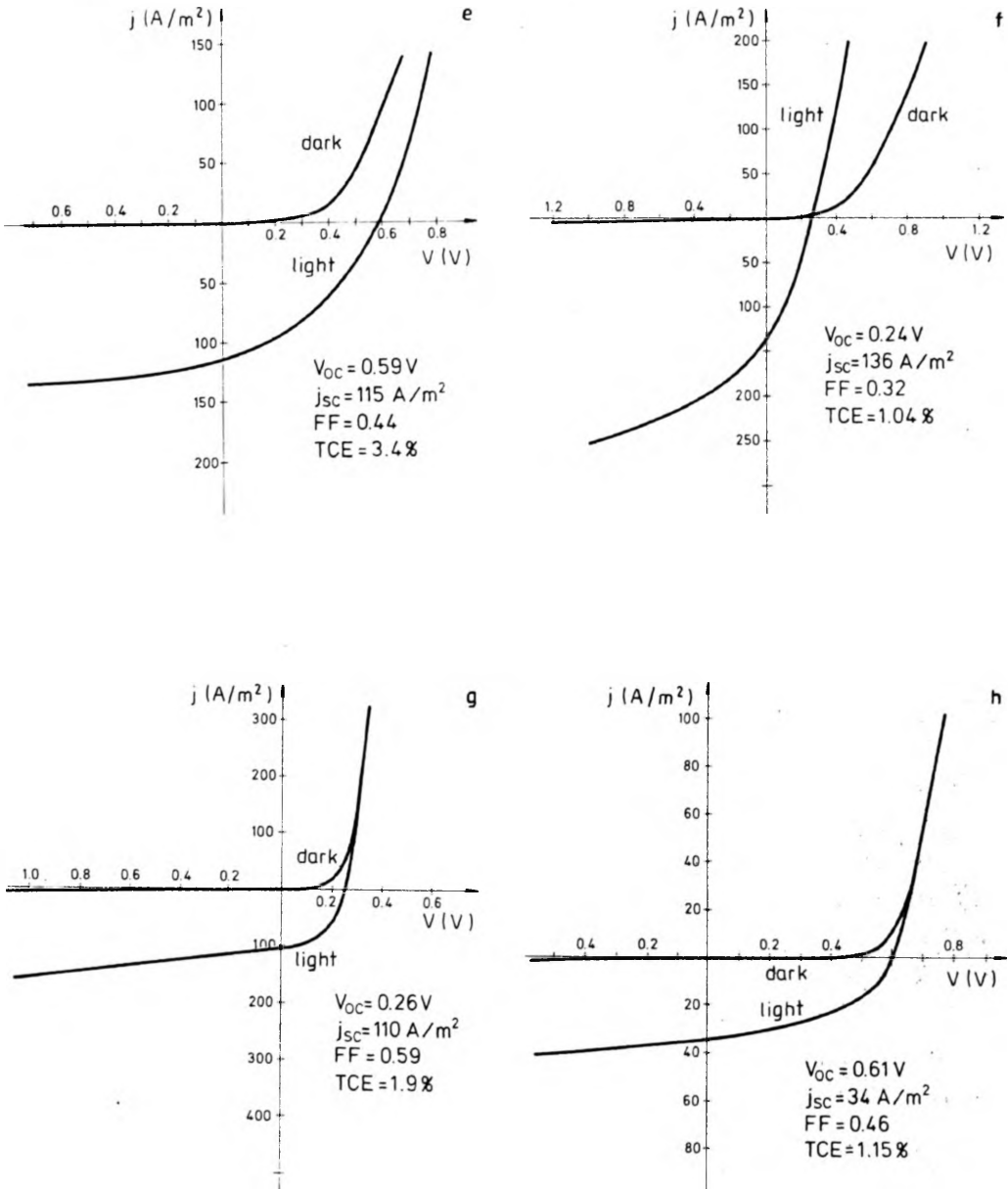
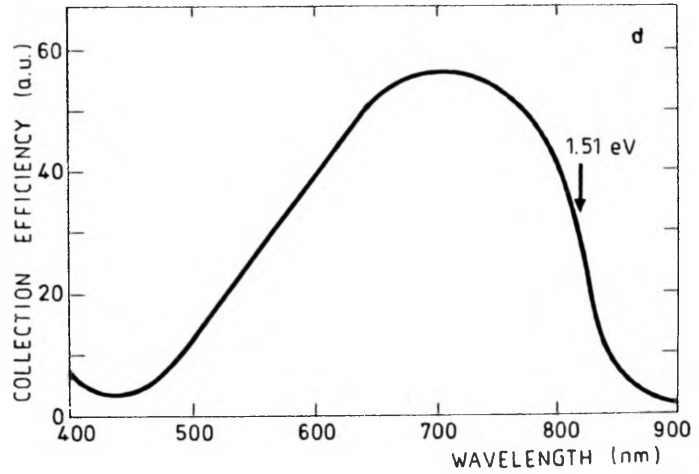
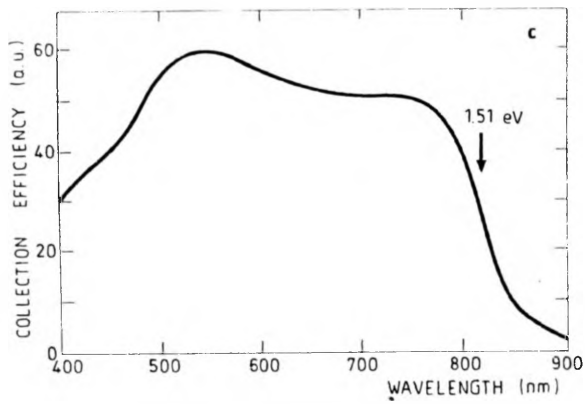
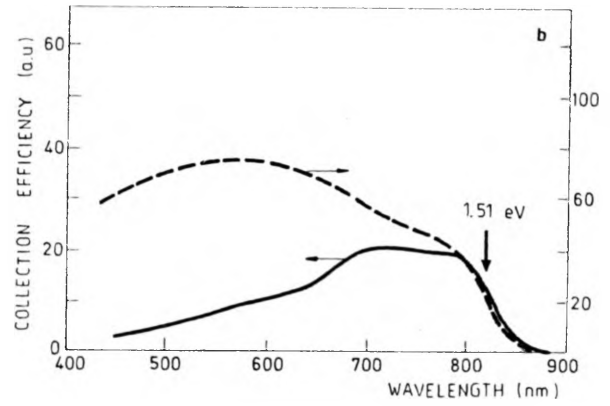
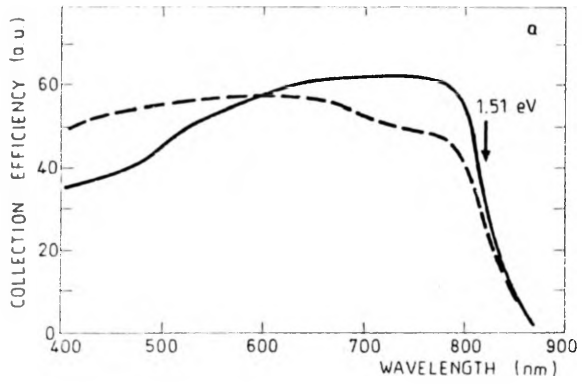


Fig. 5. Current voltage characteristics of prototype Zn<sub>3</sub>P<sub>2</sub>-based solar cells: **a** – OMG Mg-Zn<sub>3</sub>P<sub>2</sub> device with no AR coating, after [34]; **b** – OMG Mg-Zn<sub>3</sub>P<sub>2</sub> device with 80 nm SiO AR coating, after [36]; **c** – TMF 15 nm Mg-Zn<sub>3</sub>P<sub>2</sub> device with 500 nm SiO AR coating, after [36]; **d** – TMF 10 nm Mg-Zn<sub>3</sub>P<sub>2</sub> device with 70 nm ZnO AR coating, after [39]; **e** – n-p Zn<sub>3</sub>P<sub>2</sub> quasi-homojunction device with no AR coating, after [42]; **f** – n-p ITO-Zn<sub>3</sub>P<sub>2</sub> heterojunction device with no AR coating, after [49]; **g** – n-p ZnO-Zn<sub>3</sub>P<sub>2</sub> heterojunction device with no AR coating, after [50]; **h** – n-p ZnSe-Zn<sub>3</sub>P<sub>2</sub> heterojunction device with no AR coating, after [43]. “Light” curves in Fig. **a-e** were measured under AM1 and these curves in Figs. **g, h** under AM1.5 conditions of illumination. Curves in Fig. **f** were measured under 1000 W/m<sup>2</sup> illumination density



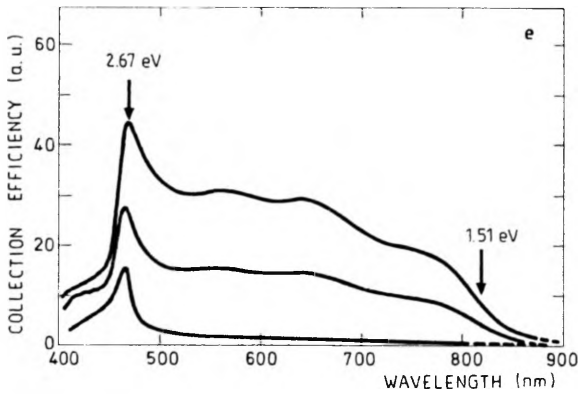


Fig. 6. Overall collection efficiency (photoresponse measured in output-current mode) of prototype Zn<sub>3</sub>P<sub>2</sub>-based solar cells: **a** – OMG Mg-Zn<sub>3</sub>P<sub>2</sub> device with no AR coating (dashed line), after [34], and that with 93 nm SiO coating (solid line), after [36], (both under AM1 conditions of illumination). **b** – TMF 10 nm Mg-Zn<sub>3</sub>P<sub>2</sub> device with 70 nm ZnO AR coating (dashed line), after [36], and n-p Zn<sub>3</sub>P<sub>2</sub> quasi-homojunction device with no AR coating (solid line), after [42], (both the curves under AM1 conditions of illumination). **c** – n-p ITO-Zn<sub>3</sub>P<sub>2</sub> heterojunction device with no AR coating, under 1000 W/m<sup>2</sup> illumination, after [49]. **d** – n-p ZnO-Zn<sub>3</sub>P<sub>2</sub> heterojunction device with no AR coating, under AM1.5 conditions of illumination, after [50]. **e** – n-p ZnSe-Zn<sub>3</sub>P<sub>2</sub> heterojunction device with no AR coating, after [43], (curves were measured under AM1.5 light conditions and voltage bias of 2 V, 1 V and 0; from top to bottom, respectively)

The maximum photocurrent (short-circuit current density) computed for Zn<sub>3</sub>P<sub>2</sub> solar cell as a function of Zn<sub>3</sub>P<sub>2</sub> thickness is shown in Fig. 4a together with the data taken from [31] for Si- and GaAs-based solar cells. The lower curve for Zn<sub>3</sub>P<sub>2</sub>-based devices is computed for bare surface (no antireflection coating) semiconductor; the upper one shows the potentiality of devices with the best multilayer antireflection coating having reflectivity factor of 5% approximately. The computations have been done under AM1 conditions of illumination (power density equal to 930 Wm<sup>-2</sup>).

Figure 4b shows the experimental current-voltage characteristics of three prototype devices under AM1.5 conditions. This figure reveals the place of the prototype Zn<sub>3</sub>P<sub>2</sub>-based solar cells in comparison to the best ones fabricated on the

Loss figures and output data of Zn<sub>3</sub>P<sub>2</sub>-based solar cells under AM1.5 illumination integrated over 0–∞ waverange, with AR coating, after [27]

Type of device	Total optical losses [%]	Total electrical losses [%]	Short-circuit current [A/m <sup>2</sup> ]	TCE [%]
OMG Schottky junction	76.9	10.7	137	12.4
TMF Schottky junction	87.7	5.1	73.5	7.2
p-n homojunction	71.5	13.3	165	15.2
p-n heterojunction	69.6	14.2	181	16.2
Bare Zn <sub>3</sub> P <sub>2</sub>	68*	14.9	190	17.1

\* value of  $\approx 60\%$  is the lowest number of optical losses assuming the best multilayer AR coating with  $R \approx 5\%$

base of Si and GaAs. The data for the Si- and GaAs-based devices have been taken from [32] and for the  $\text{Zn}_3\text{P}_2$ -based one from [27].

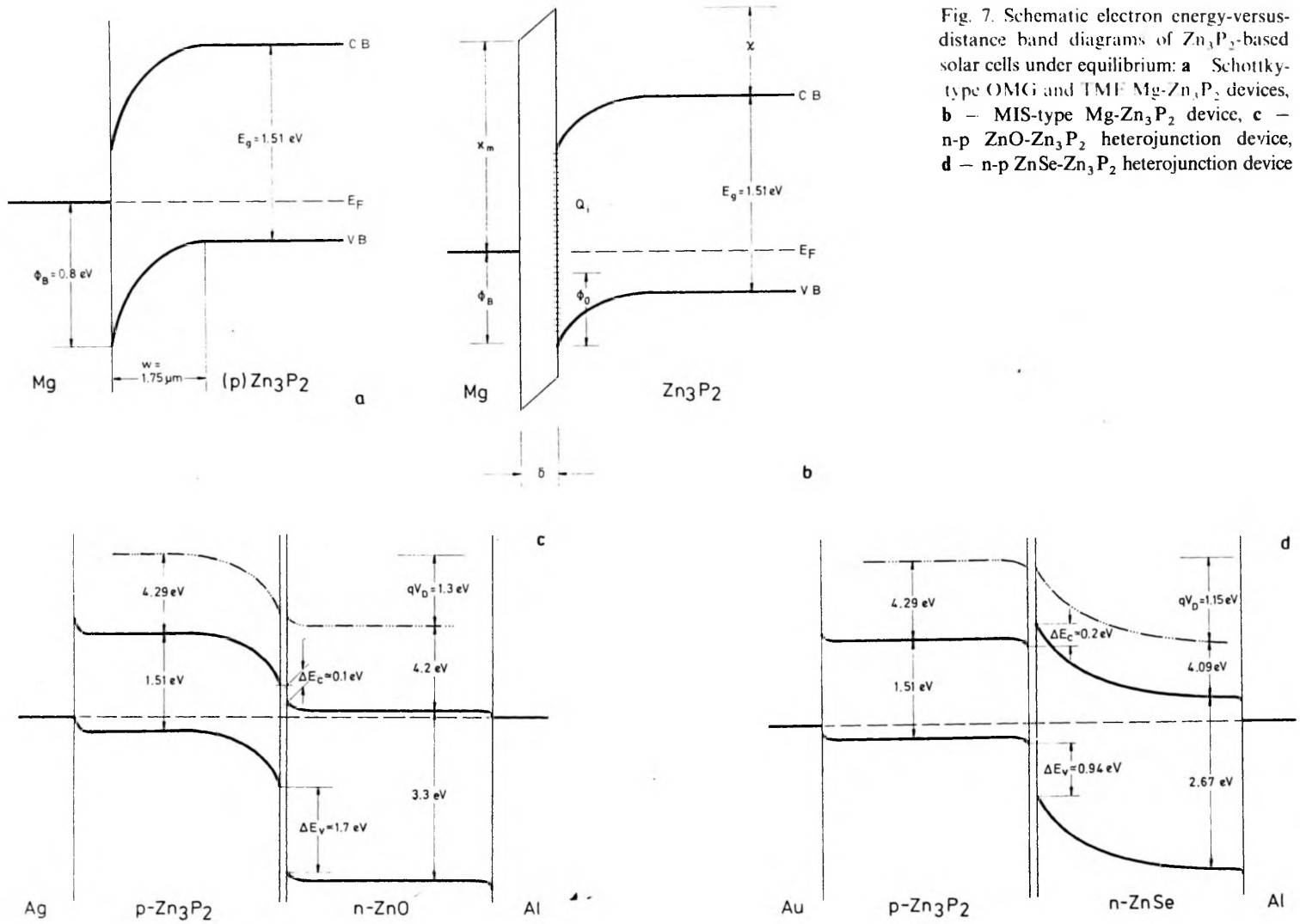
Potentiality of  $\text{Zn}_3\text{P}_2$ -based solar cells shown in Fig. 4a is comparable to that of GaAs-based devices. Reality exposed in Fig. 4b does not match the potentiality, unfortunately, for some reasons that will be discussed in details below.

## 6.2. Metal- $\text{Zn}_3\text{P}_2$ devices

Both the OMG and TMF Schottky-type metal- $\text{Zn}_3\text{P}_2$  devices have been fabricated and tested. Among all the metals used, Mg has been found to form (statistically) the highest (0.8 eV) potential barrier with  $\text{Zn}_3\text{P}_2$  (see above) although the other metals have also been examined.

*The OMG prototype solar cells* have been made of both the polycrystalline bulk samples and the polycrystalline thin film specimens [33]–[36]. The best parameters of the OMG cells have been to date presented in [34], [36]. By using magnesium grid pattern whose line separation was approximately twice the minority carrier diffusion length (and with no antireflection coating) an active area current density as high as  $j_{sc} = 183 \text{ A/m}^2$  was obtained with  $V_{oc}$  slightly lower than 0.4 V and TCE = 1.5%. For the same type OMG devices with AR coating the best parameters obtained are:  $j_{sc} = 190 \text{ A/m}^2$ ,  $V_{oc} = 0.5 \text{ V}$  and TCE up to 6% [36]. The experimental current-voltage and spectral characteristics of the best OMG-type devices without and with AR coating taken from [34] and [36], respectively, are shown in Figs. 5a, b and 6a, respectively. Dark and light  $I$ - $V$  curves intersect for these devices which can suggest a change of the transport mechanism of carriers across the junction under illumination. The reasonably flat spectral response of the devices at short wavelengths can indicate negligible losses due to surface recombination. Schematic Schottky-type equilibrium band diagram of Mg- $\text{Zn}_3\text{P}_2$  structure is shown in Fig. 7a. The band-bending region width is estimated (from simple estimation for parabolic barrier) to be 1.75  $\mu\text{m}$ .

*The TMF prototype solar cells* have been fabricated by means of dc sputtering or PVD thermal evaporation of thin (10–15 nm) magnesium layer on the top of both the polycrystalline bulk samples and the polycrystalline thin film specimens [29], [35]–[39]. The best parameters obtained for the TMF devices have been to date presented in [38], [39] for devices with 70 nm ZnS AR coating. Short-circuit current up to 178  $\text{A/m}^2$  and  $V_{oc}$  as high as 0.51 V have been attained with TCE around 4–5%. The experimental current-voltage and spectral characteristics of the best TMF-type devices (which differ in the thickness of both the Mg film and the AR coating) taken from [36], [39] are shown in Figs. 5c, d and Fig. 6b, respectively. High optical losses due to reflection from and absorption within the Mg film limit the output photo-current. The relatively low values of fill factor (not exceeding 0.6) are due mainly to the high diode factor ( $\approx 2$ ) connected presumably with the space-charge recombination mechanism. The  $V_{oc}$  value, on the other hand, is limited by the  $\approx 0.8 \text{ eV}$  maximum barrier height of the metal- $\text{Zn}_3\text{P}_2$  junction (see also above). The collection efficiency is fairly constant in the short-wavelength range. The



comparison between the devices made of the large-grain polycrystalline bulk samples and these made of the small-grain thin films shows that the grain boundaries in the large-area devices do not seriously compromise the device performance. However, we should recall that the grain-boundaries detrimental effect is always decreased under illumination [40].

Although n-type  $\text{Zn}_3\text{P}_2$  has never been prepared in any direct way and in any form (bulk or thin-film specimen)  $\text{Zn}_3\text{P}_2$  p-n homojunction formation has been reported [14], [41] in the TMF Mg- $\text{Zn}_3\text{P}_2$  devices after their heating for a few hours at 75–100°C. The evidences presented in these papers to add strength to suggestion of the p-n junction creation can, however, be (in part, at least) ascribed to the concept of the dipole-layer formation (see also Sect 5). This metal-dipole layer-semiconductor structure is, in general sense, similar to the MIS one shown in Fig. 7 b. The formation of the interface between Mg and  $\text{Zn}_3\text{P}_2$  and particularly its thickness  $\delta$  is obviously time- and temperature-dependent as well. Nevertheless, the experimental current-voltage and spectral characteristics of nominally p-n  $\text{Zn}_3\text{P}_2$  homojunction devices are shown in Fig. 5e and Fig. 6b, respectively, after [42]. The values of  $j_{sc}$  up to 153 A/m<sup>2</sup> and these of  $V_{oc}$  as high as 0.57 V have been measured under AM1 illumination conditions. The FF value in the range of 0.4–0.47 has been found giving TCE values up to 3.7%. Therefore, the overall performance of the TMF Mg- $\text{Zn}_3\text{P}_2$  devices is lower after heating, i.e., after suggested transformation into p-n homojunction devices. Particularly, the decrease in  $j_{sc}$  values and the increase in  $V_{oc}$  values strengthens the concept of dipole layer formation rather than the suggestion of p-n junction creation [14], [41]. Nevertheless, the formation of n-on-p junction by heating the Mg- $\text{Zn}_3\text{P}_2$  structure is a continuously open question.

### 6.3. Heterojunctions n-p devices

The highest value of the open-circuit voltage in the prototype OMG and TMF Mg- $\text{Zn}_3\text{P}_2$  devices achieved to date (in TMF solar cells) has been  $V_{oc} \approx 0.5$  V. This relatively low value is, however, equal to the upper limit of  $V_{oc}$  estimated for Mg- $\text{Zn}_3\text{P}_2$  devices, approximately (see above). For this reason, the search for an appropriate heterojunction partner for  $\text{Zn}_3\text{P}_2$  has been carried out. The heterojunction-partner semiconductor with  $E_g > E_g(\text{Zn}_3\text{P}_2)$  is improving the  $V_{oc}$ , while that with  $E_g < E_g(\text{Zn}_3\text{P}_2)$  is giving the higher  $j_{sc}$  values; assuming that the generation and collection of minority carriers are taking place equally in both the materials, approximately. The following parameters are usually instrumental in the material selection for heterojunctions: energy bandgap; electron affinity, lattice constant and thermal expansion coefficient.

Both isostructure and anisostructure heterojunctions with  $\text{Zn}_3\text{P}_2$  have been investigated (see [43]–[45], for short review) including II–V binary and ternary compounds such as  $(\text{Cd}_x\text{Zn}_{1-x})_3\text{P}_2$  and  $\text{Mg}_3\text{P}_2$ , ITO, and II–VI binary and ternary compounds such as CdO, CdS, ZnO, ZnS, ZnSe and  $\text{Zn}_x\text{Cd}_{1-x}\text{S}$ .

*II–V binary compounds* and their solid solutions (ternary mixed crystals) are apparent isostructure partners for  $\text{Zn}_3\text{P}_2$ . However, all (but one,  $\text{Mg}_3\text{P}_2$ ) have the

energy gap smaller than that of Zn<sub>3</sub>P<sub>2</sub>. The  $E_g$  value of Mg<sub>3</sub>P<sub>2</sub> is (direct bandgap) 2.5 eV, approximately, but the material is highly unstable in air [46]. Looking for the lower bandgap n-type material to improve  $j_{sc}$  output, it appears that only Cd<sub>3</sub>P<sub>2</sub> (either  $E_g = 0.53$  eV [47] or 0.56 eV [48] at 300 K) is available. It can produce ternary mixed crystals Cd<sub>3</sub>P<sub>2</sub>-Zn<sub>3</sub>P<sub>2</sub> possessing the n-type conductivity together with the appropriate energy bandgap [36], [48]. The (n)Cd<sub>3x</sub>Zn<sub>3-3x</sub>P<sub>2</sub>-(p)Zn<sub>3</sub>P<sub>2</sub> heterojunctions with  $x > 0.5$  were investigated and the p-n barrier height was found to be 0.77 eV [44]. However, low values of both the  $V_{oc}$  and  $j_{sc}$  were measured and no further work was performed.

Heavily doped n-type and optically transparent ( $E_g = 3.6$  eV [55]) *indium-tin oxide (ITO)* seems to be an attractive partner to any p-type semiconductor. ITO-on-Zn<sub>3</sub>P<sub>2</sub> heterojunction devices have been fabricated by rf sputter deposition of about 300 nm thick ITO (having  $\approx 90\%$  of optical transmission) onto sputter cleaned Zn<sub>3</sub>P<sub>2</sub>:Ag [49]. The  $V_{oc}$  values of 0.24–0.32 V and  $j_{sc}$  values of 130–260 A/m<sup>2</sup> have been reported with total conversion efficiency up to 1.1% under 1000 W/m<sup>2</sup> illumination density. One should note the unbelievably high density (260 A/m<sup>2</sup>) of photocurrent listed in [49]. The experimental current-voltage and spectral characteristics of ITO-Zn<sub>3</sub>P<sub>2</sub> prototype solar cell are shown in Fig. 5f and Fig. 6c, respectively, after [49]. The shape of the  $I$ - $V$  curves has been ascribed in [49] to the tunneling mode as dominating transport mechanism. The very low value of FF has been attributed to high series resistance and large diode factor. The first reason is the hardly expected one in ITO-contained devices. The relatively low  $V_{oc}$  might come from carrier recombination by the interface states and/or by Zn<sub>3</sub>P<sub>2</sub> surface states. The last suggestion is strengthened by the shape of the spectral characteristic shown in Fig. 6c, in which the high drop of photocurrent in blue region is well visible.

*II-VI binary compounds* such as thoroughly studied CdS, ZnS, ZnO and ZnSe appear attractive candidates because (among other things) a wide variety of preparation techniques. Furthermore, they satisfy the requirements mentioned above quite well (see, e.g. [43], for short review). A natural partner for any zinc compound is zinc oxide (common cation) and (n)ZnO-(p)Zn<sub>3</sub>P<sub>2</sub> heterojunction solar cells were prepared and described in [50], [51]. The 100 nm thick low resistivity ZnO film was dc sputter deposited on sputter cleaned silver-doped polycrystalline Zn<sub>3</sub>P<sub>2</sub> wafers. The maximum values of  $V_{oc} = 0.38$  V and of  $j_{sc} = 46$  A/m<sup>2</sup> were measured under 850 W/m<sup>2</sup> illumination density (somewhere in between AM1 and AM1.5 conditions) on as-prepared devices. Low  $j_{sc}$  value is due principally to the high series resistance of ZnO and in order to reduce that resistance and to provide good ohmic contact to ZnO film the aluminium (100 nm thick) layer dots were deposited by PVD-thermal evaporation. The experimental current-voltage and spectral characteristics of the best example of such a device are shown in Fig. 5g and Fig. 6d, respectively. The best value (corrected for an active area) of  $j_{sc} = 110$  A/m<sup>2</sup> was obtained, however, a long with  $V_{oc}$  decreasing to  $V_{oc} = 0.26$  V. Fill factor up to 0.586 and TCE as high as 1.9% was found. Well visible maximum of collection efficiency arises from the p-side of heterojunction. The next peak (from ZnO region) is expected in the blue region (below 375 nm) for high ZnO energy bandgap ( $E_g = 3.3$  eV is reported in [52]).

Between these peaks there is a rapid fall-off, suggesting strong recombination at the interface. The schematic equilibrium band diagram of n-p ZnO-Zn<sub>3</sub>P<sub>2</sub> heterojunction is shown in Fig. 7c. It has been constructed using the known values of electron affinities and energy band gaps [43]. Note the differences between the model shown in Fig. 7c and the one presented in [50] as Fig. 4. Diffusion voltage as high as  $V_D = 1.3$  V is deduced from the diagram. It appears likely that the interface states are highly instrumental in decreasing the  $V_{oc}$  with reference to the relatively high  $V_D$  value.

Two others II–VI compounds, namely n-type CdS ( $E_g = 2.42$  eV) and n-type ZnS ( $E_g = 3.6$  eV), have provided even noticeably worse results than those for ZnO [43], [44]. High leakage current and low values of both the  $V_D$  and  $V_{oc}$  were found.

Looking for solar-spectrum transparent and reasonably low electron affinity material, n-type thin ZnSe film ( $E_g = 2.67$  eV [56] at 300 K) appears to be a promising common-cation partner. The (n)ZnSe:Ga-(p)Zn<sub>3</sub>P<sub>2</sub> heterojunctions were prepared, essentially by the thin-film technique described in [43], [56]. The experimental results of current-voltage and spectral characteristics measured under AM1.5 illumination conditions are shown in Fig. 5h and Fig. 6e, respectively. The values of  $V_{oc}$  as high as 0.84 V and  $j_{sc}$  up to 37.5 A/m<sup>2</sup> were obtained. Fill factor slightly lower than 0.5 along with the TCE value up to 1.8% were found. Low value of  $j_{sc}$  (in comparison with the estimated one listed in the Table, particularly) is (most probably) the result of the high resistivity of the ZnSe film and then high series resistance of the device. Equilibrium schematic energy band diagram of (n)ZnSe-(p)Zn<sub>3</sub>P<sub>2</sub> heterojunction is shown in Fig. 7d after [43]. Taking the known values of parameters of both the semiconductors [43], the value of  $V_D = 1.15$  V is deduced from the diagram, being in a reasonable reference to the measured  $V_{oc}$  values (much better than for ZnO-Zn<sub>3</sub>P<sub>2</sub> heterojunction – see above). As it results from Fig. 7d, almost the entire built-in-field region is hidden inside the n-type semiconductor. This is in favour of higher photo-current output from ZnSe than that from Zn<sub>3</sub>P<sub>2</sub> as shown in Fig. 6e (note that this situation is quite different than that in the case of both the ITO-Zn<sub>3</sub>P<sub>2</sub> and the ZnO-Zn<sub>3</sub>P<sub>2</sub> heterojunctions). Carrier collection in Zn<sub>3</sub>P<sub>2</sub> is then affected by generation-recombination mechanism in the bulk with almost no barrier electric field involved. At least, two order of magnitude higher donor concentration in ZnSe is needed to optimise the heterojunction [43]. However, obtaining of highly doped n-type ZnSe films has been an apparent problem [53], [54], [56]. The impact of the recombination at the heterojunction interface upon the  $j_{sc}$  value is presumably high as well.

## 7. Final conclusions

Zn<sub>3</sub>P<sub>2</sub> has become one of the promising semiconductors for applications in photovoltaic solar energy conversion. Both the values of the direct energy bandgap and the diffusion length of electrons promote the reasonably high energy conversion efficiency factor. Also, both Zn and P are abundant and inexpensive. Prototype Zn<sub>3</sub>P<sub>2</sub>-based solar cell devices in various configurations have been recently



fabricated and tested. It has been shown that the prototype devices meet roughly the design calculation results and that the value of TCE about 6% can be accomplished relatively easily. After some optimization procedure, TCE up to 16% can be achieved [27] strengthening the need to promote further investigations of both the fundamental properties of Zn<sub>3</sub>P<sub>2</sub> and the electronic performance of Zn<sub>3</sub>P<sub>2</sub>-based photovoltaic thin film devices.

Investigations of the nature of the photoelectric properties of Zn<sub>3</sub>P<sub>2</sub> and Zn<sub>3</sub>P<sub>2</sub>-based devices still leave many unanswered questions. One of it is the presence of deep levels within Zn<sub>3</sub>P<sub>2</sub> bandgap since they are probably responsible for the carrier tunneling under intense illuminations, eventually resulting in low  $V_{oc}$  values. Also, the surface properties of Zn<sub>3</sub>P<sub>2</sub> are of an unusually great interest, being furthermore closely related to the device performance.

As far as prototype solar cells are concerned, all the empirical results suggest a dominant role of the interface (between Zn<sub>3</sub>P<sub>2</sub> and its partner to form a potential barrier). An exception is p-n homojunction device and one of the evident courses of action is to fabricate the p-n junction in Zn<sub>3</sub>P<sub>2</sub>. However, the problem of the appropriate doping of Zn<sub>3</sub>P<sub>2</sub> to convert it into n-type material has to be solved above all. On the other hand, magnesium appears to be the only metal that is capable of providing beneficial interface reaction to Zn<sub>3</sub>P<sub>2</sub>. OMG and TMF Mg-Zn<sub>3</sub>P<sub>2</sub> junctions are therefore the reasonably good configurations for solar cell applications, providing that suitable fabrication procedure is used. The low  $V_{oc}$  value limited by the potential barrier height of Mg-Zn<sub>3</sub>P<sub>2</sub> structure to  $V_{oc} \approx 0.5$  V is however a distinct disadvantage of the Mg-Zn<sub>3</sub>P<sub>2</sub> devices.

Heterojunction-type Zn<sub>3</sub>P<sub>2</sub>-based solar cells are of a great interest having a clear advantage of possibility to obtain very high  $V_{oc}$  in case of wide-bandgap partner material. Such a heterojunction partner can also serve as an optical window/antireflection coating. ZnSe appears to be very appropriate semiconductor on condition that the successful high ( $\approx 10^{21} \text{m}^{-3}$ ) donor doping will be assured, and electronic properties of the metallurgical interface ZnSe-Zn<sub>3</sub>P<sub>2</sub> will be improved, as well.

In any case of configuration, all-thin-film solar cell structures are required to achieve the best conversion efficiency and the ultimate economic goal for a large-scale electricity generation.

*Acknowledgements* – The fruitful discussions with Dr Dr M. Bhushan, A. W. Catalano, and J. D. Meakin from IEC, University of Delaware; and Dr Dr N. Mirowska, J. Misiewicz and K. Nauka from IP, Technical University of Wrocław, are gratefully acknowledged.

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Received May 4, 1988

### Солнечные батареи на базе $Zn_3P_2$

Представлен обзор актуального состояния исследований солнечных батарей на базе  $Zn_3P_2$ . Настоящая работа содержит краткое описание основных свойств  $Zn_3P_2$ , обсуждение некоторых технологических проблем, а также теоретическую оценку параметров солнечных батарей с  $Zn_3P_2$ . Обсуждены также самые лучшие достижения прототипных батарей в конфигурации соединения Шотки (OMG и TMF) и гетеросоединения.